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Note: Not all of the reactions will occur. For those that do not, write no reaction. 1. A piece of copper is dropped into a container of water. No reaction 2. Liquid bromine is added to a container of sodium iodide crystals.  $\text{Br}_2(l) + 2\text{NaI}(s) \rightarrow 2\text{NaBr}(s) + \text{I}_2(s)$  3. An aluminum strip is immersed in a solution of silver nitrate.

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## Quiz on all Chemical Reactions - Max Study

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ammonium oxalate 4 224 polonium(VI) thiocyanate  $\text{Po}(\text{SCN})_6$  tetraphosphorus decaoxide  $\text{P}_4\text{O}_{10}$   
12. 13. 15. 16. 17. 18. 19. manganese(VII) oxide copper(II) dihydrogen phosphate

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### **KEY to 5 Reaction Prediction - Ipscience.com**

The Ultimate Chemical Equations Handbook Determine the oxidation number of the underlined element:  $\text{KMnO}_4$  Since K is an alkali metal, its charge must be 1+ Oxygen is 2— but there are four of them, therefore, 4 times 2— equals 8— If 1+ and 8— are added Ultimate Chemical Equation Answers

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